



Corporate Office : Aakash Tower, 8, Pusa Road, New Delhi-110005. Phone : (011) 47623456

Aakash National Talent Hunt Exam 2021

SAMPLE PAPER

(Class XI Studying Moving to Class XII)

Time : 1 Hour

MM : 90

ANSWERS

| Physics | Chemistry | Mathematics |
|----------------|----------------|---------------|
| 1. (3) | 12. (2) | 23. (4) |
| 2. (4) | 13. (4) | 24. (4) |
| 3. (1) | 14. (1) | 25. (3) |
| 4. (1) | 15. (3) | 26. (3) |
| 5. (2) | 16. (4) | 27. (2) |
| 6. (1, 3, 4) | 17. (1, 2) | 28. (2) |
| 7. (1, 3) | 18. (2, 3) | 29. (1, 2, 3) |
| 8. (1) | 19. (2) | 30. (3, 4) |
| 9. (4) | 20. (2) | 31. (2, 3) |
| 10. (3) | 21. (1) | 32. (3) |
| 11. A → (P, S) | 22. A → (R, S) | 33. (1) |
| B → (R) | B → (P, S) | 34. (1) |
| C → (P, Q) | C → (Q, R, S) | 35. A → (Q) |
| D → (P) | D → (P, Q, S) | B → (R) |
| | | C → (P) |
| | | D → (S) |



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ANSWERS & SOLUTIONS

PHYSICS

1. Answer (3)

If $A = B + C$, then

$$[A] = [B] = [C]$$

2. Answer (4)

$$a = \frac{d^2s}{dt^2} = \frac{d}{dt}(4t + 3t^2) = (4 + 6t) \text{ m/s}^2$$

At $t = 1$ s,

$$a = 10 \text{ m/s}^2$$

3. Answer (1)

For maximum range,

$$\theta = \frac{(90 - 30)}{2}$$

$$\therefore \theta = 30^\circ$$

4. Answer (1)

$$a_1 = a_2 = \frac{5g \sin 37^\circ - 2g}{(5+2)} = \left(\frac{g}{7}\right) = \frac{10}{7} \text{ m/s}^2$$

5. Answer (2)

$$W_T = \Delta KE$$

$$\Rightarrow W_F + W_{mg} = 0$$

$$\Rightarrow W_F = mgh$$

6. Answer (1, 3, 4)

$$F_x \leq \mu N$$

$$\Rightarrow 10 \leq \mu \cdot 40$$

$$\Rightarrow \frac{1}{4} < \mu$$

7. Answer (1, 3)

At $X = X_{\max}$,

$$v_1 = v_2 = \frac{v_0}{2}$$

$$\Delta KE = \frac{1}{2}mv_0^2 - 2 \cdot \frac{1}{2}m\left(\frac{v_0}{2}\right)^2 = \frac{mv_0^2}{4}$$

8. Answer (1)

9. Answer (4)

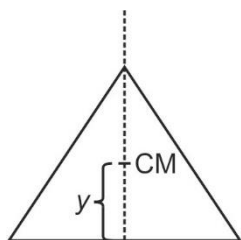
Solution for Q. Nos. 8 and 9 :

$$a_p = a_0 - R\alpha \quad (\text{Rolling condition})$$

10. Answer (3)

Work done by conservative forces may be positive or negative.

11. Answer A(P, S); B(R); C(P, Q); D(P)



$$y = \frac{L\sqrt{3}}{2} \cdot \frac{1}{3} = \frac{L}{2\sqrt{3}}$$

CHEMISTRY

12. Answer (2)

Bond order \propto bond dissociation energy

$$\text{Bond order } (\text{O}_2)^{2-} = 1.0$$

$$\text{Bond order } (\text{O}_2)^- = 1.5$$

$$\text{Bond order } (\text{O}_2) = 2.0$$

$$\text{Bond order } (\text{O}_2^+) = 2.5$$

13. Answer (4)

For isoelectronic ions,

$$IE \propto \text{Atomic number } (Z)$$

14. Answer (1)

For complete neutralisation,

$$100 \times M = 50 \times \frac{1}{2}$$

$$\Rightarrow M = \frac{1}{4} \text{ molar}$$

15. Answer (3)

$$\text{Number of moles of He + unknown gas} = \frac{5.7 \times 4}{0.08 \times 300} = 0.95$$

$$\text{Number of moles of unknown gas} = 0.95 - 0.76 = 0.19$$

$$\frac{r_{\text{He}}}{r_x} = \sqrt{\frac{M_x}{4}} = \frac{0.76}{0.19} = 4; \quad M_x = 64$$

16. Answer (4)

$$C_p = C_v + R = 25 + 0.03T + R$$

$$dQ = nC_p dT = 2(25 + 0.03T + R) dT$$

$$dS = \frac{dQ}{T} = 2(25 + 0.03T + R) \frac{dT}{T}$$

$$\Delta S = 50 \ln \frac{T_2}{T_1} + 0.06(T_2 - T_1) + 2R \ln \frac{T_2}{T_1}$$

$$\Delta S = 50 \ln \frac{600}{300} + 0.06(600 - 300) + 2 \times 8.314 \ln \frac{600}{300} = 64.17 \text{ JK}^{-1}$$

17. Answer (1, 2)

$$\text{Binding energy of 4th state} = \frac{13.6Z^2}{(n)^2} = 13.6 \text{ eV}$$

$$\Rightarrow Z = 4$$

$$\text{Binding energy of 2nd excited state} = \frac{13.6(4)^2}{(3)^2} = 24.17 \text{ eV}$$

18. Answer (2, 3)



$$\text{Milliequivalents of H}_2\text{O}_2 = 25 \times 2 \times 0.5 = 25$$

$$\text{Milliequivalents of KMnO}_4 = 50 \times 5 \times 0.2 = 50$$

$$\text{Milliequivalents of KMnO}_4 \text{ left unreacted} = 50 - 25 = 25$$

$$\text{Moles of KMnO}_4 \text{ left unreacted} = \frac{25}{5 \times 1000} = 0.005$$

$$\text{Moles of O}_2 \text{ produced} = \frac{25}{2 \times 1000} = 0.0125$$

19. Answer (2)

$$\text{Density of the gas, } d = 2p - 0.2p^2$$

$$\text{Density will be maximum, when } \frac{d(d)}{dp} = 0$$

$$\frac{d(d)}{dp} = 2 - 0.4p = 0; \quad p = 5 \text{ atm}$$

20. Answer (2)

$$d_{\text{max}} = \frac{M}{\text{Molar volume}}; \quad \text{Molar volume} = \frac{M}{5} \text{ litre}$$

21. Answer (1)

$$n\text{-Factor of } H_3PO_4 = 2$$

$$\text{Equivalent weight of } H_3PO_4 = \frac{98}{2} = 49$$

22. Answer A(R, S); B(P, S); C(Q, R, S); D(P, Q, S)

$$n = 6 \longrightarrow n = 3; \quad 6 \text{ lines in the IR region}$$

$$n = 7 \longrightarrow n = 3; \quad 10 \text{ lines in the IR region}$$

$$n = 5 \longrightarrow n = 2; \quad 3 \text{ lines in the visible region and 3 lines in the IR region}$$

$$n = 6 \longrightarrow n = 2; \quad 4 \text{ lines in the visible region and 6 lines in the IR region}$$

MATHEMATICS

23. Answer (4)

$$\cos 2\theta = 2 \cos^2\theta - 1$$

$$= 1 - 2 \sin^2\theta$$

$$= \cos^2\theta - \sin^2\theta$$

$$\text{But } 2 \sin\theta \cdot \cos\theta = \sin 2\theta$$

24. Answer (4)

$$x^2 - y^2 + 2xyi + x - iy = 0$$

$$\Rightarrow x^2 + x - y^2 = 0 \quad \text{and} \quad 2xy - y = 0$$

$$y = 0 \quad \text{or} \quad x = \frac{1}{2}$$

$$y = 0; \quad x = 0, -1$$

$$x = \frac{1}{2}, \quad y = \pm \frac{\sqrt{3}}{2}$$

The possible values of z are

$$z = 0 + 0i, \quad -1 + 0i, \quad \frac{1}{2} + \frac{\sqrt{3}}{2}i \quad \text{and} \quad \frac{1}{2} - \frac{\sqrt{3}}{2}i$$

25. Answer (3)

$$S = \frac{15}{2} \{2 + 14 \times 3\} = 330$$

26. Answer (3)

$$(2p - 3q)^2 + (3q - 4r)^2 + (4r - 2p)^2 = 0$$

$$\Rightarrow 2p = 3q = 4r$$

$$\Rightarrow \frac{1}{p}, \frac{1}{q} \text{ and } \frac{1}{r} \text{ are in AP.}$$

$$\Rightarrow p, q \text{ and } r \text{ are in HP.}$$

27. Answer (2)

$$1 + (-1) + \frac{1}{2} = \frac{1}{2}$$

28. Answer (2)

$$16 - x^2 \geq 0 \Rightarrow x \in [-4, 4]$$

29. Answer (1, 2, 3)

$$f(x) = x^2 + x$$

$$\Rightarrow f(2) = 6, f(-1) = 0, f(-2) = 2, f(3) = 2$$

30. Answer (3, 4)

$$P \cap Q = \{3, 4, 5\}$$

$$Q \cap R = \{4, 6\}$$

$$P \cap Q \cap R = \{4\}$$

31. Answer (2, 3)

$$\frac{9}{8} + 2 \cos^2 x + \cos x - 1 = 2 \cos^2 x + \cos x + \frac{1}{8} \quad (D \geq 0)$$

32. Answer (3)

MASHU EE NN

Total words – (words in which EE together + NN together – EE and NN both together)

$$= \frac{9!}{2!2!} - \left(\frac{2 \cdot 8!}{2!} - 7! \right) = 11(7!)$$

33. Answer (1)

EEAU, MNNSH

Select only 4 positions for vowels and arrange in 1 way.

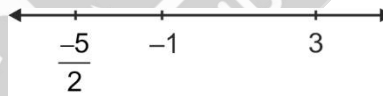
$$\therefore \text{Total number of arrangement} = {}^9C_4$$

34. Answer (1)

$$(x^2 - 2x - 3)(2x + 5) \leq 0$$

$$(x + 1)(x - 3)(2x + 5) \leq 0$$

$$x \in \left(-\infty, -\frac{5}{2} \right] \cup [-1, 3]$$



35. Answer A(Q); B(R); C(P); D(S)

$$(A) S_n = \frac{1}{3} - \frac{1}{4} + \frac{1}{4} - \frac{1}{5} + \dots + \frac{1}{n+2} - \frac{1}{n+3} \Rightarrow S_n = \frac{n}{3(n+3)}$$

$$\therefore S_\infty = \frac{1}{3}$$

$$(B) S_\infty = \frac{1}{1 - \frac{2}{3}} = 3$$

$$(C) HM = \frac{2\alpha\beta}{\alpha + \beta} = \frac{2(8 + 2\sqrt{5})}{(4 + \sqrt{5})} = 4 \Rightarrow \frac{1}{H} = \frac{1}{4}$$

$$(D) (G.M.)^2 = \frac{1}{9} \cdot \frac{1}{4}$$

$$G.M. = \frac{1}{6}$$

